

Perovskite Quantum Dots Luminescent

Nanofluids as Biological Thermometer by

Ratiometric Method



Dr. Juan Jesús Gallardo¹, Ana Jiménez¹, Dr. Eduardo Blanco² and Dr. Javier Navas¹

¹Physical Chemistry Department, University of Cádiz. Spain

²Physics of Condensed Matter, University of Cádiz. Spain

jj.gallardo@uca.es

Outline...



Introduction







Introduction...

- The measure of temperature is an important parameter in our environment, specially in biological media.
- For measuring *in vivo*, the development of methodologies that can permit the determination of changes in temperature, at cell scale, with good sensibility is a need.
- One strategy is the use of *nanothermometry*, where the luminescent emission of nanoparticles depends with temperature, and can be used as thermometer at the nanoscale.
- The ideal nanothermometer must meet a series of premises such as excitation and emission within the biological windows (NIR), emission shift with temperature within the range 10-50 °C, good photo and chemical stability and biocompatibility.

Introduction...

- For the determination of temperature, a *ratiometric method* can be applied.
- This method uses the linear relation between the <u>intensity ratio</u> of two emission bands of the nanoparticles at the excitation wavelength, and the <u>inverse of temperature</u>. So a good nanoparticle candidate should have two emission bands at the excitation wavelength.

$$\operatorname{og} \frac{I_{\lambda_1}}{I_{\lambda_2}} = a + b\left(\frac{1}{T}\right)$$

• Good candidates could be nanoparticles of quantum dots and nanomaterials doped with rare earths in a core-shell structure, where the core are the quantum dots, and the shell is formed by the second material.



Introduction...

- This work is a starting point, representing first steps to the development of nanothermometer.
- Synthesis of 'core' nanoparticles with luminescent properties and preparation of nanofluid in non polar media



Perovskite structure



Luminescent nanofluid

Experimental...

- Synthesis of CsPbBr₃, CsPbBr_{1.5}I_{1.5} and CsPbI₃ nanocrystals by <u>hot-injection method</u>
 - 1. Preparation of cesium oleate solution (Cs_2CO_3 + oleic acid + octadecene)
 - 2. Preparation of lead halide solution (PbX₂ + ODE + oleylamine + oleic acid)
 - 3. Injection of cesium oleate over lead halide solution at 150 $^{\circ}$ C
 - 4. After 5 seconds, immersion of the flask in ice bath to stop the reaction
 - 5. Centrifugation of the crude at 0 ^oC. Nanocrystals are obtained as precipited
- Preparation of CsPbBr₃, CsPbBr_{1.5}I_{1.5} and CsPbI₃ nanofluids
 - 1. Dispersion of nanocrystals in anhydrous toluene.

Experimental...

- Characterization
- 1. Powder X-Ray Diffraction (XRD)
- 2. High Resolution Transmission Microscopy (HR-TEM)
- 3. Energy Dispersive X-ray Spectroscopy (EDS)
- 4. Photoluminescent Spectroscopy (PL)



• Powder X-Ray Diffraction (XRD)



• High Resolution Transmission Microscopy (HR-TEM) / Energy Dispersive X-ray Spectroscopy (EDS)



LESS THAN 20 nm

• Photoluminescent Spectroscopy (PL)



• Ratiometric method: Relation between emission and temperature

$$\log \frac{I_{\lambda_1}}{I_{\lambda_2}} = a + b\left(\frac{1}{T}\right)$$



Conclusions...

- Three perovskites nanocrystals (CsPbBr $_{1-x}I_x$) have been synthesized by hot-injection method.
- Three nanofluids with the perovskites nanocrystals have been prepared.
- All nanofluid are luminescent with two emission band.
- A shift of emission to higher or lower wavelength with temperature in function of composition has been recorded.

Conclusions...

- A ratiometric method has been applied to establish a relation between band's emissions and temperature.
- $CsPbBr_{1.5}I_{1.5}$ based nanofluid shows the best results.
- Perovskite nanocrystals are good candidates as core particles in a core-shell structure.
- This work establishes a starting point to achieve a nanothermometer with application in biological media.



Thank you

for your

attention



Dr. Juan Jesús Gallardo

Physical Chemistry Department, University of Cádiz. Spain

jj.gallardo@uca.es